



**Date :** Wednesday 29 October 2008  
Mercredi 29 octobre 2008

**Title/Titre :** A new method to measure oxygen ingress into a wine bottle

**Speaker/  
Intervenant :** Elizabeth Waters (The Australian Wine Research Institute,  
Australia)





# 14 closures = 14 different wines

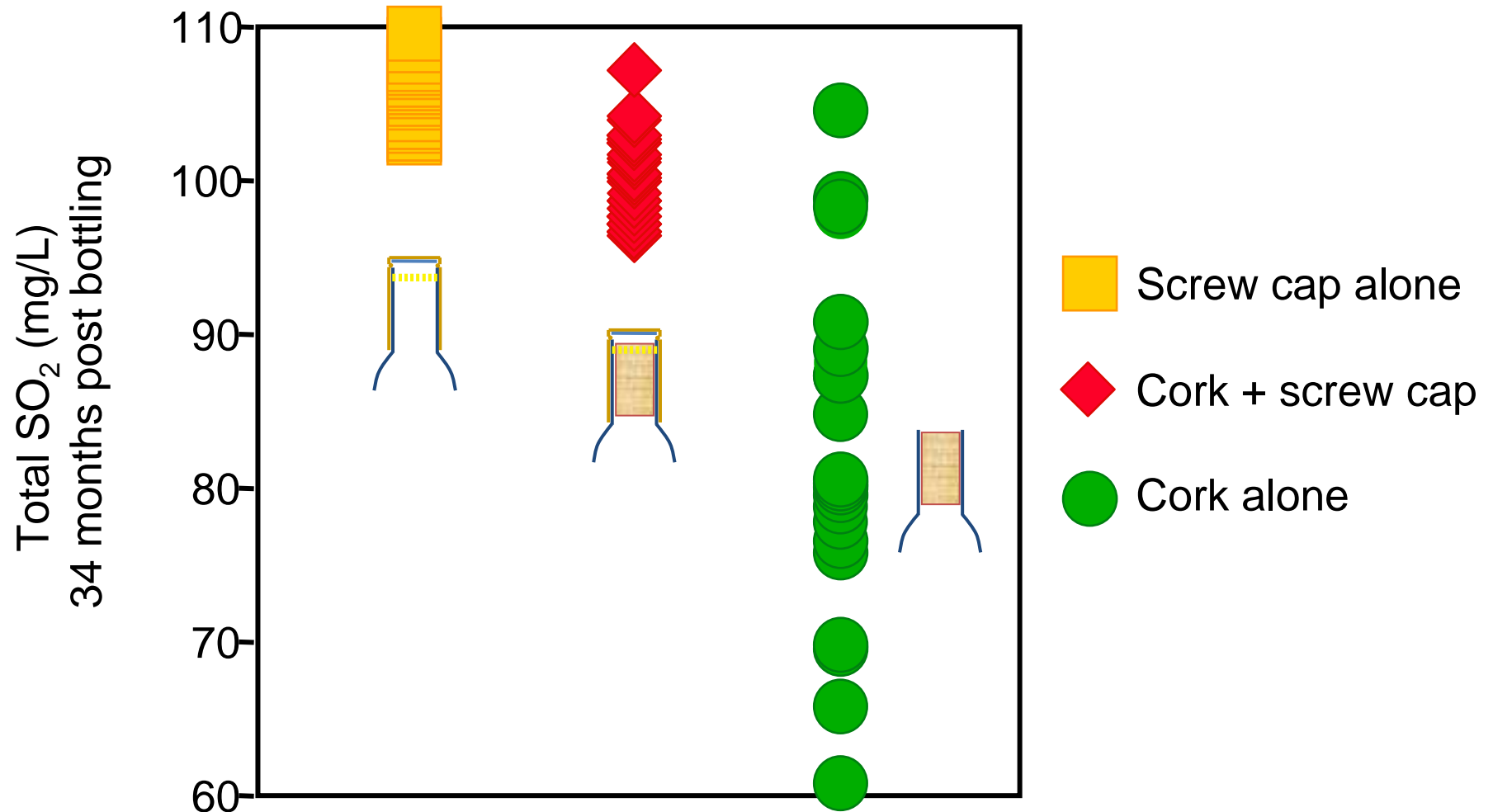


Range of colour 63 months post bottling (Godden et al 2001 Aust. J. Grape Wine Res. 7, 62-105 )





# Variation in OTR



(Caloghiris et al 1997 Aust. J. Grape Wine Res. 3:9-17)



# How to determine OTRs ?

New method developed at AWRI

Uses two reagents in model wine

- **methylene blue** and **light** to convert oxygen from triplet to singlet state
- bis-9,10-anthracene-(4-trimethylphenylammonium) dichloride (**BPAA**) as an oxygen trap

BPAA is coloured

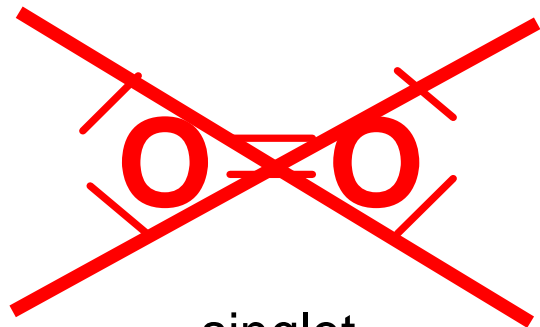
Reaction between oxygen and BPAA

- produces a colourless product
- is stoichiometric
- consumes oxygen

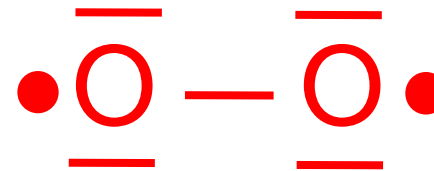




# What does an oxygen molecule look like?

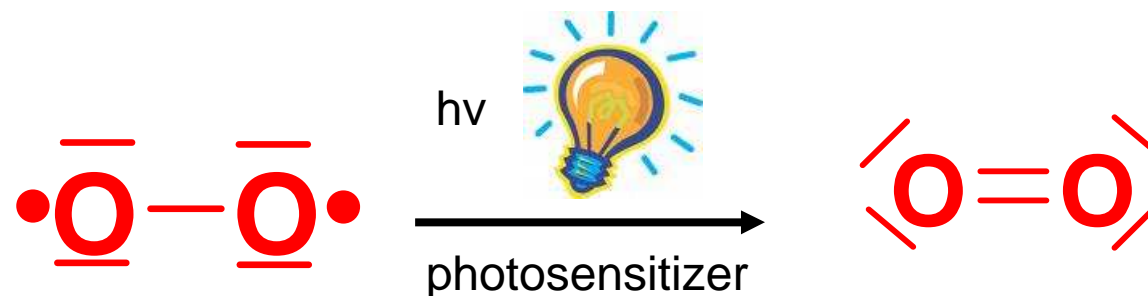


singlet



triplet

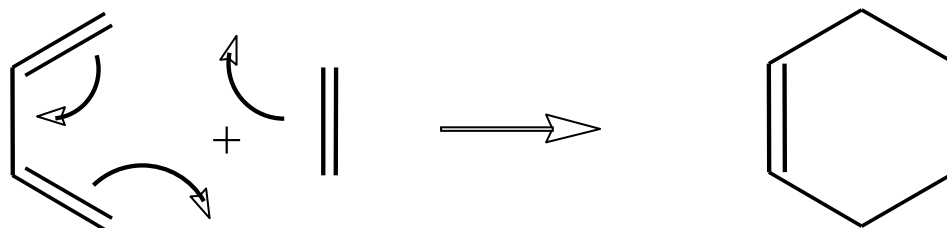
- 
- Excited oxygen: singlet state.



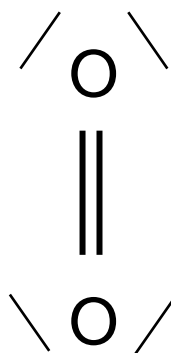


# Diels Alder Reaction

Diels-Alder reactions utilize a diene (4  $\pi$  electrons) and a dienophile (2  $\pi$  electrons)



Dienophile is singlet oxygen





# Criteria for a diene singlet oxygen trap

**Soluble in wine-like medium: 10 -15% (v/v) ethanol in water, pH 2.8 - 4.2**

- more than 1900 compounds reported as singlet oxygen traps
- Most are organic i.e. not water soluble
- Most of the water soluble conjugated dienes are insoluble in acidic medium.

**The first published water soluble cation singlet oxygen trap was described in 1997: BPAA**

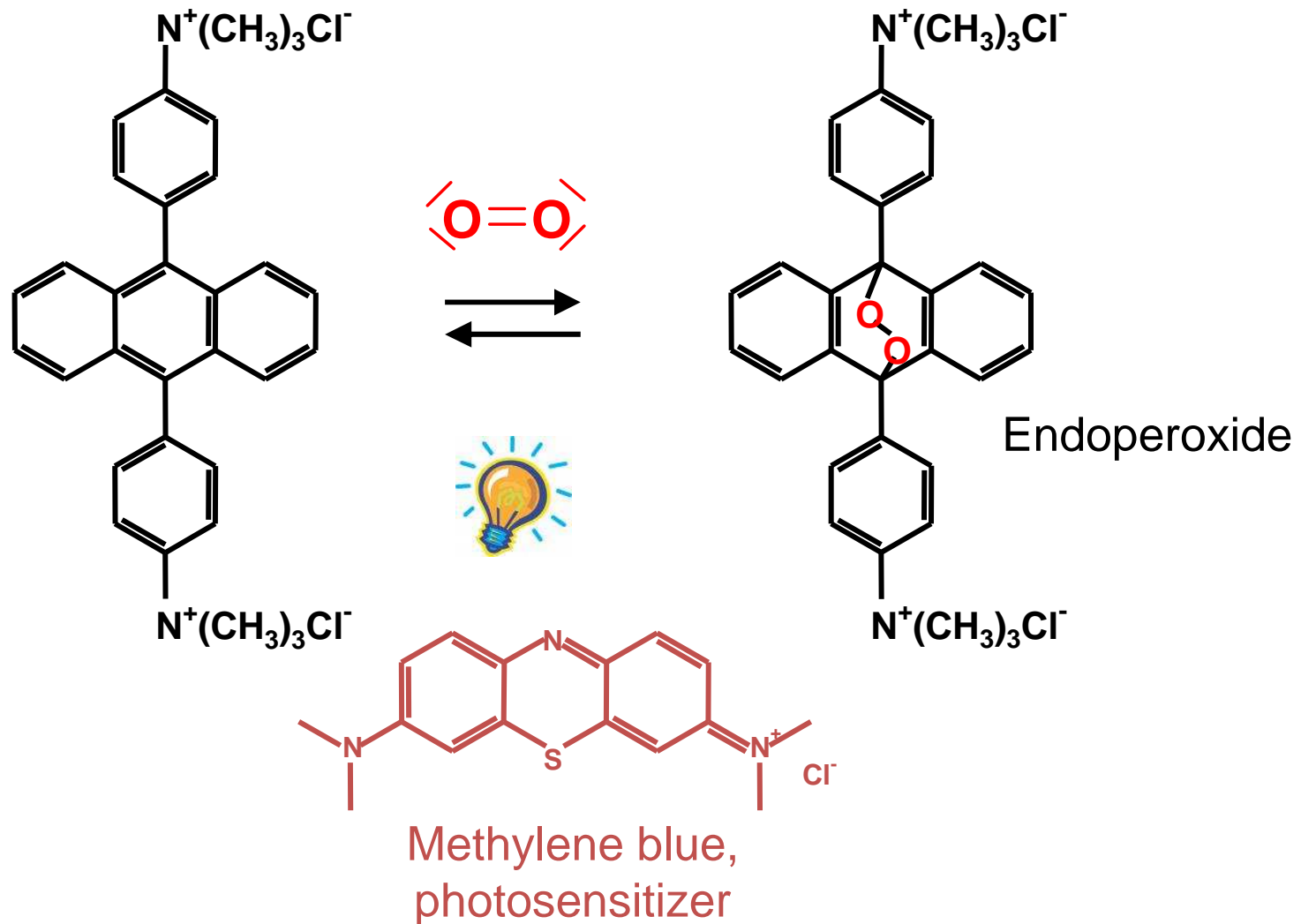
**No reaction with the photosensitizer**

**Have visible absorption**





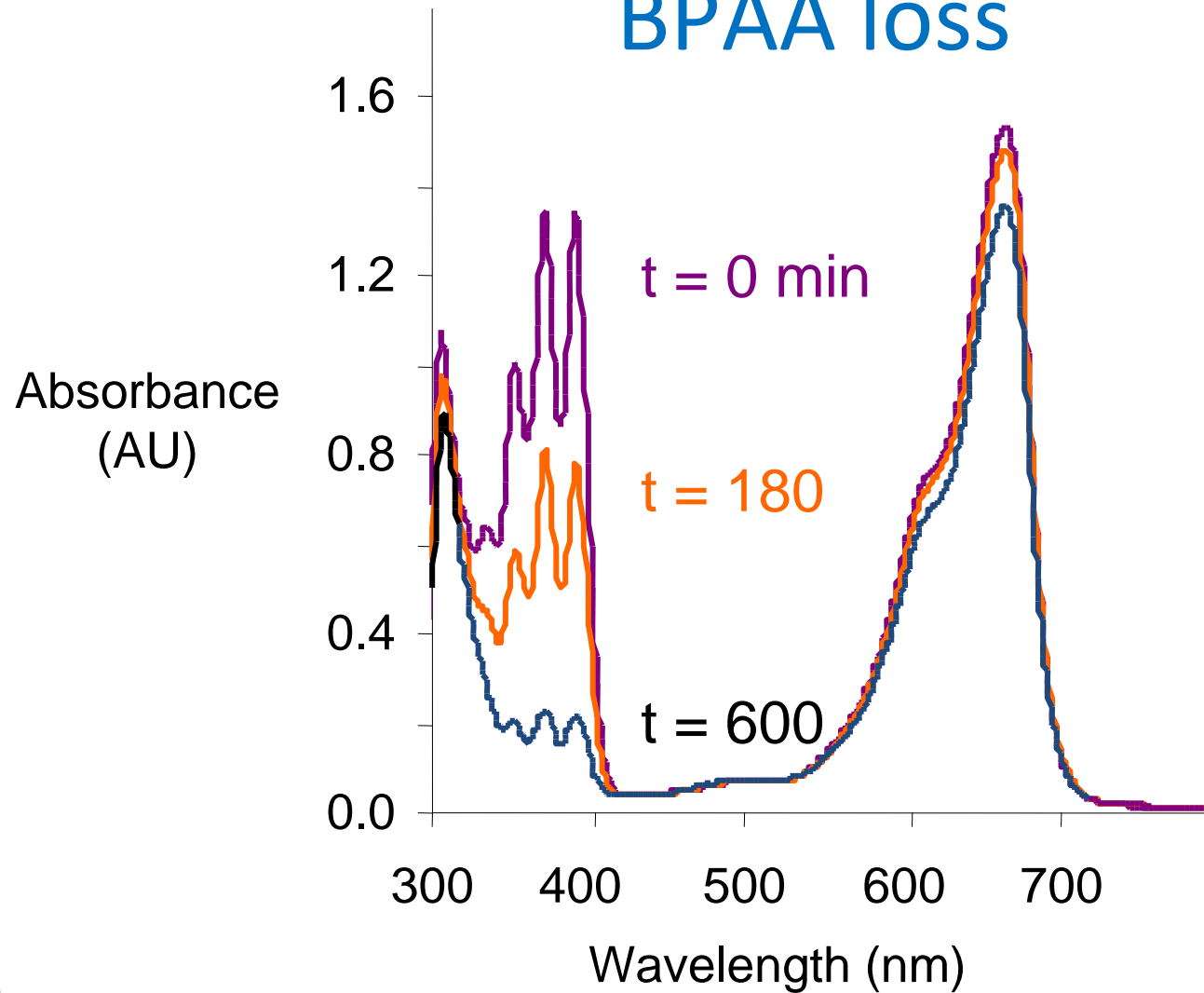
## BPAA: Bis-9,10-anthracene-(4-trimethylphenylammonium) dichloride





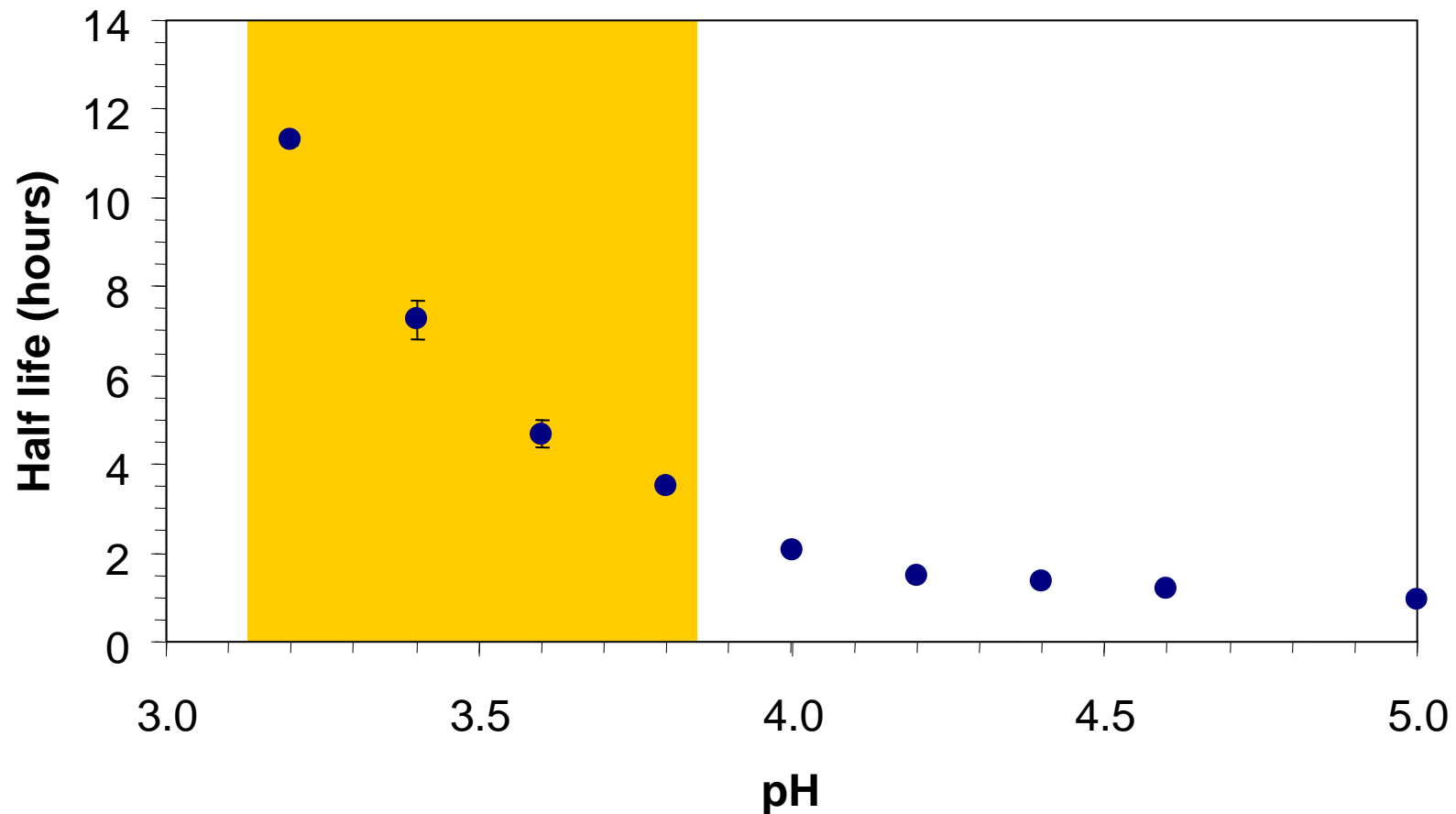


# Monitoring oxygen ingress by BPAA loss





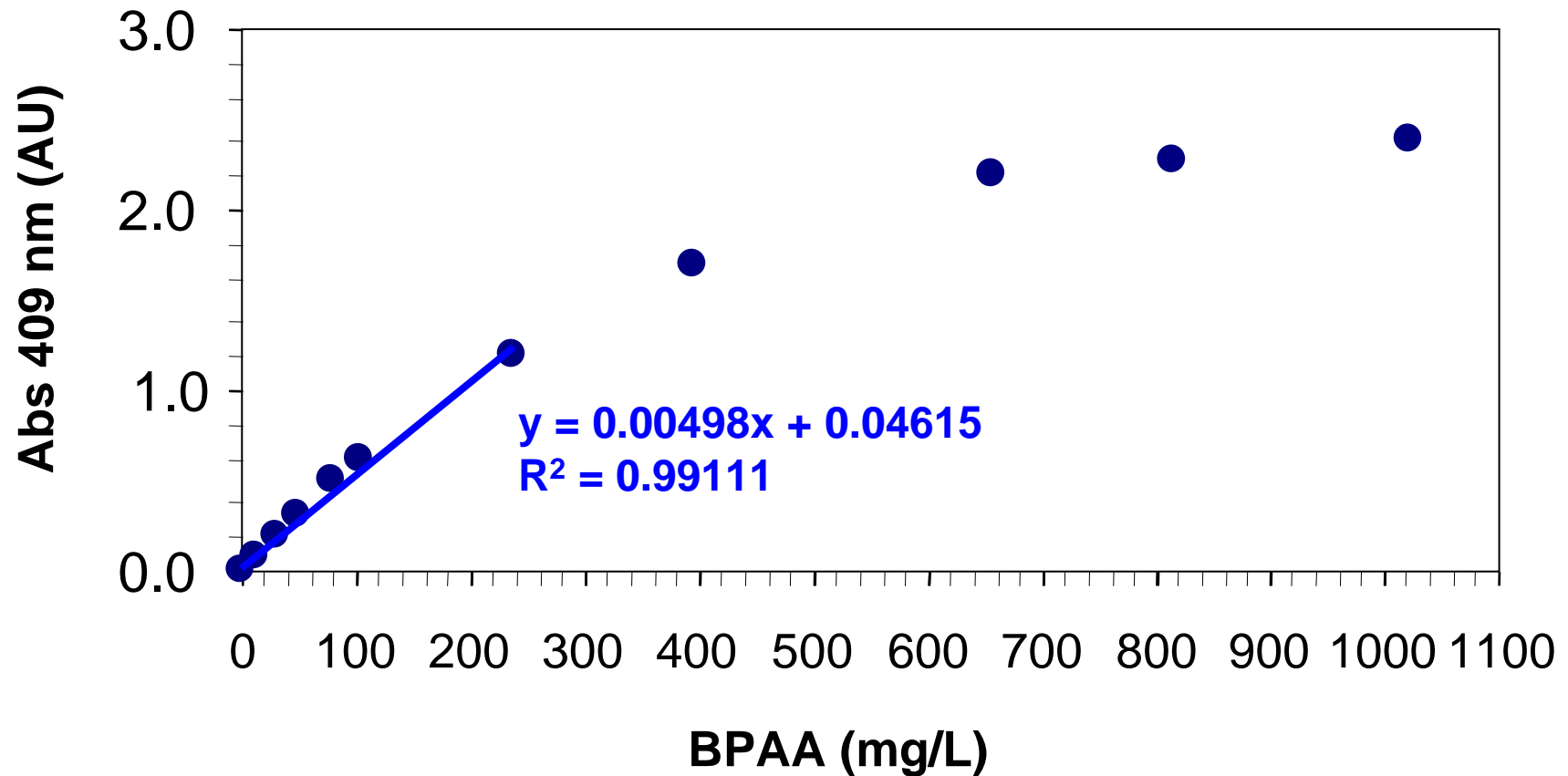
# Impact of pH on BPAA half life



Conditions: acidified (HCl) 10% ethanolic solution open to the air

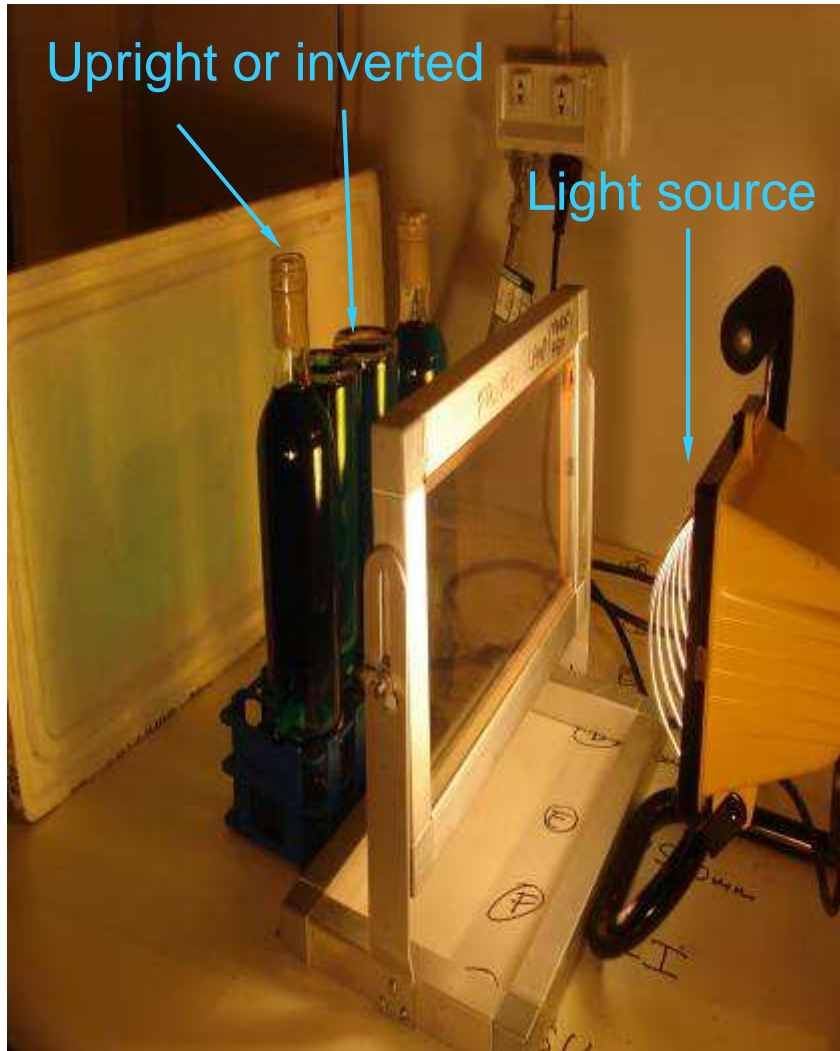


# Determining BPAA concentration in bottle



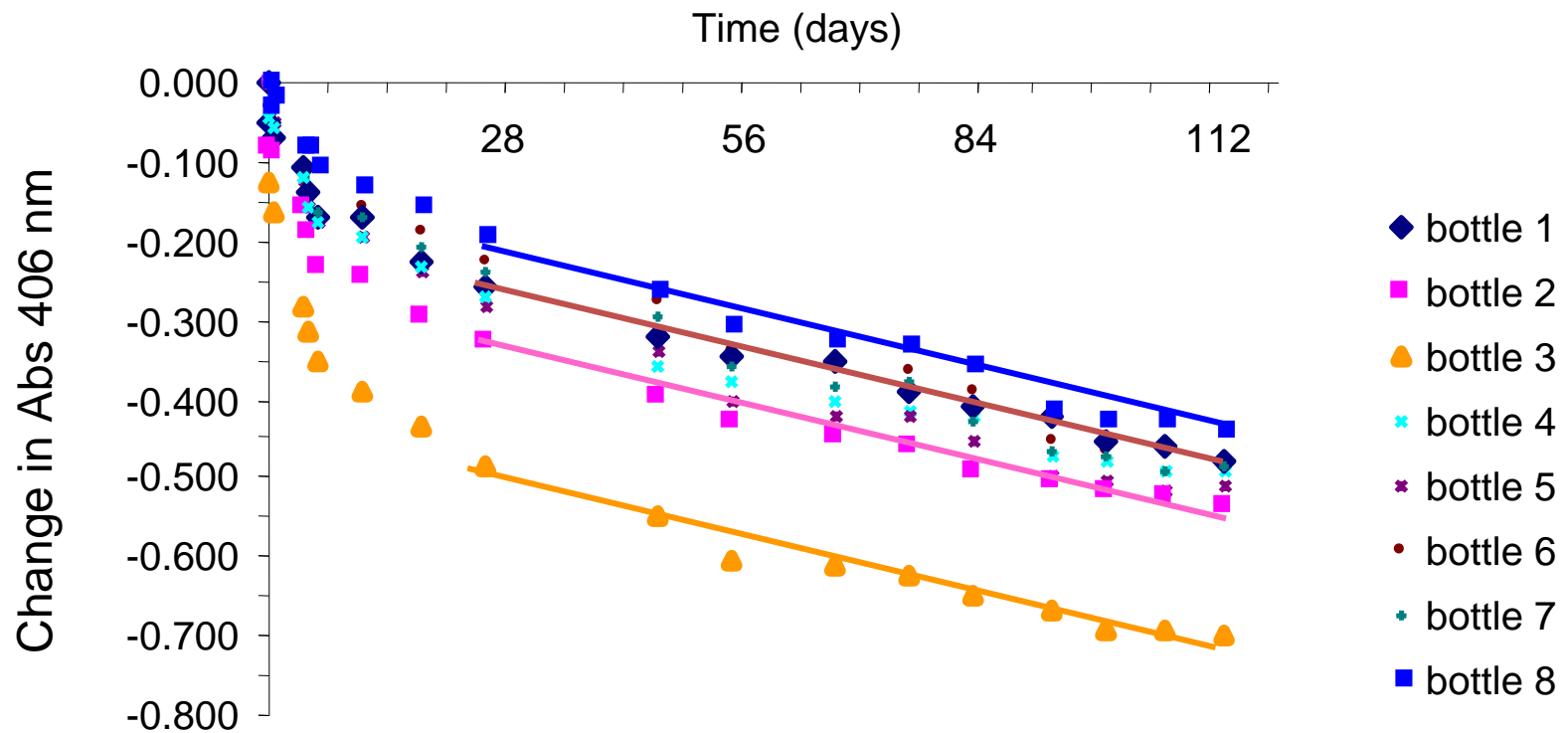


# Simple set up



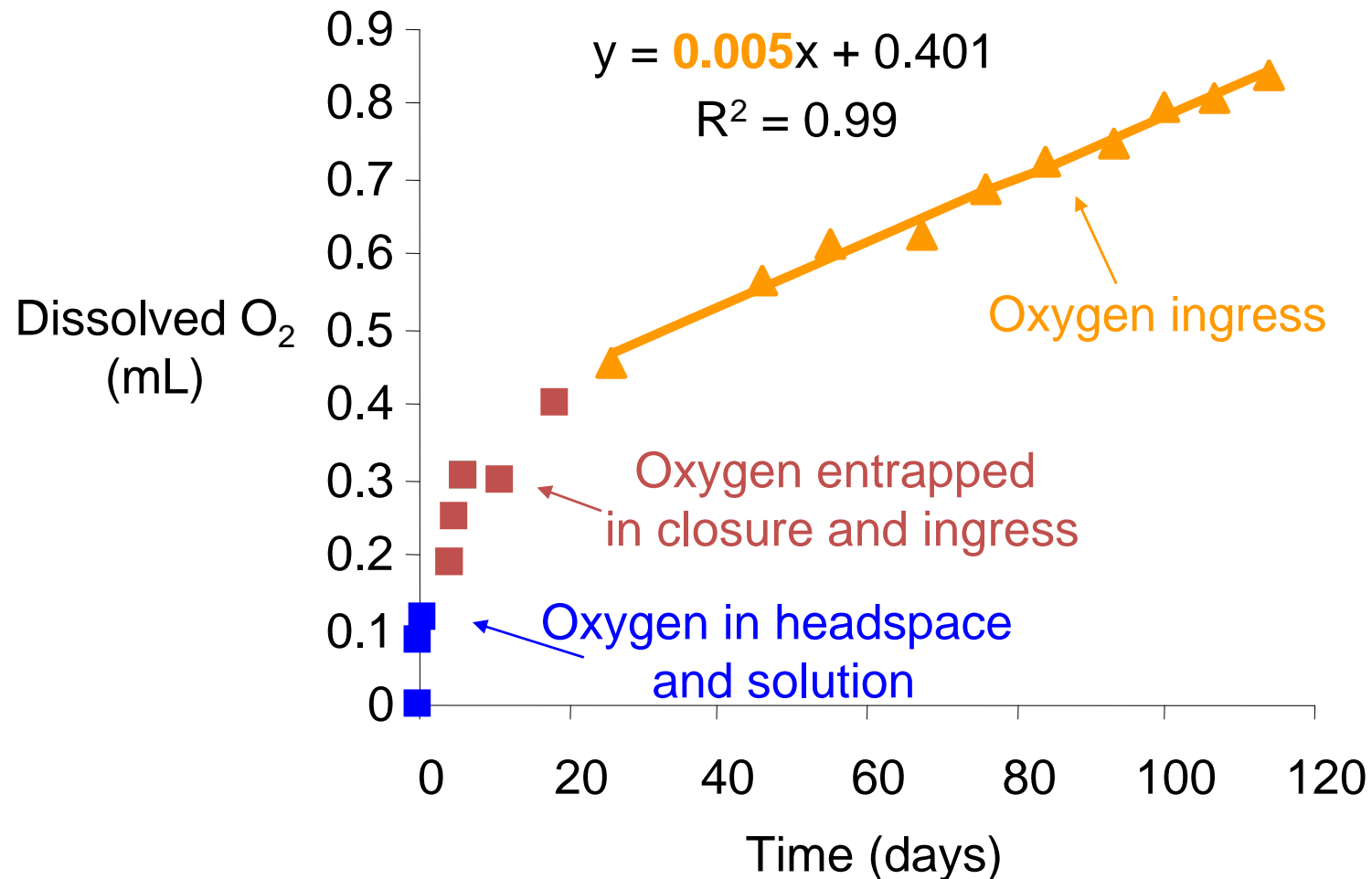


# Example of raw data obtained





# Interpreting the data





## How does the BPAA method compare to others?

For a synthetic closure, upright storage

### Mocon

- **0.0050** mL O<sub>2</sub> / day

Lopes et al (2005) J Agric Food Chem.  
53: 6967-6973

- **0.0078 - 0.009** mL O<sub>2</sub> / day

### BPAA

- **0.0052** mL O<sub>2</sub> / day.



Picture courtesy of Jim Peck,  
G3 Enterprises





# Pros and cons

## Disadvantages

- Only measuring dissolved oxygen
- Cannot be done in real wine

## Advantages

- Model wine mimics real life conditions (humidity, ethanol)
- Oxygen is consumed
- Also provides data on oxygen pickup at bottling







# Summary

## BPAA method can estimate

- Oxygen permeation rates **PLUS**
- Oxygen entrapped in the closure
- Oxygen pickup at bottling

## This tool allows wine producers to

- Select the most appropriate closure for the wine
- Audit bottling line performance





# Acknowledgements

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